AQA GCSE Physics Energy & Power Calculations Worksheet - ANSWERS

1. The Jumping Frog

 Answer: Gravitational potential energy = mass * gravitational field strength * height = 0.1 kg * 9.8 N/kg * 0.5 m = 0.49 J

2. The Hot Bath

• **Answer:** Change in thermal energy = mass * specific heat capacity * temperature change = 100 kg * 4200 J/kg°C * (45°C - 40°C) = **2,100,000 J**

3. The Speedy Car

• Answer: Kinetic energy = 0.5 * mass * (speed)² = 0.5 * 1000 kg * (20 m/s)² = 200,000 J

4. The Stretched Spring

Answer: Elastic potential energy = 0.5 * spring constant * (extension)² = 0.5 * 200 N/m * (0.1 m)² = 1 J

5. The Falling Rock

Answer: Gravitational potential energy at the top = kinetic energy just before impact (assuming no energy loss) mass * gravitational field strength * height = 0.5 * mass * (speed)²
9.8 N/kg * 50 m = 0.5 * (speed)² speed² = 980 speed = √980 = 31.3 m/s (approximately)

6. The Cooling Cup

• **Answer:** Change in thermal energy = mass * specific heat capacity * temperature change = 0.25 kg * 4200 J/kg°C * (20°C - 80°C) = -63,000 J (The negative sign indicates energy is lost)

7. The Lifting Weightlifter

Answer: Work done = force x distance = mass * gravitational field strength * distance = 150 kg * 9.8 N/kg * 2 m = 2940 J Power = work done / time = 2940 J / 3 s = 980 W

8. The Efficient Light Bulb

• Answer: Time in seconds = 5 hours * 60 minutes/hour * 60 seconds/minute = 18,000 s Energy transferred = power * time = 10 W * 18,000 s = **180,000 J**

9. The Rolling Ball (Challenge)

- Answer a): Energy lost = Initial gravitational potential energy Final kinetic energy = 10 J 7 J = 3 J
- Answer b): Power = work done / time = (change in gravitational potential energy) / time = 10 J / 2 s = 5 W

10. The Kettle (Challenge)

• Answer: Energy required to heat water = mass * specific heat capacity * temperature change = 1.5 kg * 4200 J/kg°C * (100°C - 20°C) = 504,000 J Time = energy transferred / power = 504,000 J / 2500 W = 201.6 seconds (approximately) In reality, it would take longer because

some energy is always lost to the surroundings (e.g., heating the kettle itself, heat loss to the air). This means the kettle is not 100% efficient.